LIFE & LEGACY OF HERMAN SCHLUNDT

Schlundt risked health for research

SCHLUNDT from page **1A**

The making of a physical chemist Schlundt was born in 1869 in Two Rivers, Wis.,

and completed his undergraduate work at the University of Wisconsin. In 1899, he traveled to the University of Leipzig

in Germany to work in the laboratory of Wilhelm Ostwald, a pioneering physical chemist. According to science historian John W. Servos, Ostwald is often listed with Swedish physicist-turned-chemist Svante Arrhenius and Dutch chemist Jacobus Henricus van 't Hoff, Jr. as a father of physical chemistry

In his lab, Ostwald taught his students to apply the principles of physics, such as energy, to basic units of matter. In addition to Schlundt, more than 40 Americans were drawn to his lab from the 1890s to the early 1900s.

As a German-American from a Midwestern town, Schlundt differed from most of these chemists, who came from English families in the Northeast and Middle Atlantic.

After studying in Leipzig, most of this elite group went on to become chemistry professors at top private universities in the U.S. Others became leading government and industrial chemists. Schlundt was one of the few who decided to teach

at a public, land-grant university. After earning his Ph.D. from the University of Leipzig in 1901. Schlundt came to MU. At that time, radioactivity was emerging as a subject for intense scientific study, as well as a popular sci-

ence fad. Schlundt kept a collection of pamphlets and newspaper articles throughout his career that advertised pseudo-scientific gimmicks to those seeking miracle cures. The collection provides a glimpse into the radiation hype of the early 20th Century.

'Miracle cure'

Radium, for example, was touted as a marvel of modern science, able to heal everything from cancer to eczema. "Radium - The Magic Element!" trumpeted one pamphlet that advertised a device called the Revigerator that was hyped as a way to irradiate drinking water.

An ad for a similar product, the Georadium Drinking Apparatus, proclaimed: "To-day it is universally conceded that the field of radium therapy at once is the most intriguing, the most mysterious and potentially the most valuable scientific development in the history of human achievement."

A news item declared that radium was capable, if properly employed, of preventing people from

growing old Schlundt became fascinated with radioactive metals, which appear ordinary on the surface but do emit powerful and mysterious energy. In 1904, he began meeting with another chemistry professor twice a week to discuss the emerging study of radioactive elements. They read all the papers on the sub ject they could get their

hands on. "Scarcely two months passed by before we found ourselves building electroscopes," Schlundt wrote in a 1931 letter to university president Walter Williams.

One of Schlundt's early experiments was testing water from one of the university's deep wells for radioactivity. It tested positive (and still does, though at a level below the Missouri Department of Natural Resource's maximum contaminant level). Schlundt then expanded his search for radiation to hot springs and aquifers all across the country.

stone National Park for the U.S. Geological Survey.

of radium." he wrote.

From 1906 to 1908, he tested the springs in Yellow-



PLATE NO. 2. AUTOCLAVE AND FILTER PRESS.

Courtesy of University Archives (C:6/17/1) Equipment used for extracting and refining radium from carnotite ores once stood in Pickard Hall. This photo was originally published in a 1923 MU research bulletin, written by Howard Barker and Herman Schlundt.

In fact, throughout his long tenure at MU, he received water samples from people still looking to score a scientist's endorsement of the healing power of their springs and wells. Eventually, Schlundt turned his attention to radioactive metals.

Radium from ore

He had gotten an introduction to radium refining in the summer of 1913 at the U.S. Bureau of Mines station in Denver.

The bureau was working on refining carnotite ore, a sandstone streaked with neon yellow crust mined in western Colorado and eastern Utah. This ore was a vital source of radium for researchers all across the world

> "I spent last summer in Colorado in the U.S. Bureau of Mines and came in touch with the experiments in progress there on the separation of radium from the low grade ores of western Colorado," Schlundt wrote in a letter dated 1914.

The method he learned used a series of messy, labor-intensive steps to convert a lot of ore into tinv crystals of

"Although a large quantity of ore must be worked upon to get a very small quantity of radium at present, still this ore to-day is the principal source

During Schlundt's summer in Denver, he spent time with Samuel Colville Lind, who became a lifelong friend and colleague. Lind was a decade younger than Schlundt, Tennessee-born and also an alumnus of Ostwald's laboratory in Leipzig. According to a biography of Lind by the National Academies Press, the work they did was laborious and dangerous. Lind ended up burning away half of his right thumb and index finger after years of holding radium in his bare hands. The summer they spent together provided the social connection that enabled Schlundt to get into

the refining business later.

Golden partnership

In 1914, Schlundt took the work back with him to the university, partnering with a researcher named Howard H. Barker, who had once processed ore commercially.

Schlundt persuaded several corporations and individuals to send him donations of radium ore. Altogether, he received more than four tons of raw

Radium was spread so thin throughout the ore that a ton might only yield a couple hundred milligrams, the weight of a small pill. That meant radium and other radioactive metals were more valuable than gold or diamonds in the late 1910s and early 1920s.

According to "New International Yearbook: A Compendium of the World's Progress," published in 1921, radium sold for \$115 to \$120 per milligram that year. Gold cost about \$21 per troy ounce roughly 1.09 ounces — that year, according to the National Mining Association. A milligram of gold would have been worth just 0.0007 cents.

Schlundt and Barker developed techniques to improve the efficiency and reduce the cost of extracting radium. They finished their work in 1922, the year American carnotite mining stopped. A new source of cheap radium - \$70 per milligram, according to Schlundt — had been discovered in the Belgian Congo, and American mines couldn't compete.

This development generated interest in alternative sources of radiation. Mantles for lanterns

While Schlundt and Barker were working on their radium refining method, Schlundt made contact with a man whose influence would shape the rest of his own career — Harlan S. Miner with the Welsbach Co. in New Jersey.

The company made mantles for gas lanterns, a crucial item before the rise of electricity. Miner was one of the chemists who perfected the use of another radioactive metal, thorium, which glows

in a gas flame. The process of extracting thorium from its source material, monazite sand, left tons of radioactive sludge. Miner wanted to find a way to turn this waste product into cash, and he needed a radium-refining expert to help the company profit from its waste materials. In 1917, the U.S. Bureau of Mines helped Miner locate Schlundt, most likely through Schlundt's friend Lind.

Schlundt soon realized that the process he used to extract radium from ores could easily be modified to extract an isotope of radium, radium-228, from Welsbach's industrial waste. He called this isotope "mesothorium." In one of his many letters to Miner, Schlundt described them both as "mesothorium rooters.

His radium-228 research probably contributed most to the lasting radioactivity in Pickard Hall because of the huge amount of radioactive waste he needed to synthesize it.

While the radium-228 refinery was in operation, he was receiving shipments of thousands of pounds of radioactive sludge from Welsbach and the Lindsay Light Co. in Chicago. This radioactive waste ultimately contaminated the grounds of both companies, leading to their eventual listing as EPA Superfund sites.

An excerpt from Schlundt's report on refining radium-228 from the waste materials shows how much was required to produce a tiny amount of the radioactive substance. "Up to the present time, about sixteen hundred

(1,600) pounds of the original material have been processed," Schlundt wrote. This yielded only 86 milligrams of radium-228 — or about .003 ounces, less than the weight of a sewing needle. Schlundt ended up writing a pamphlet on radium-228 that the U.S. Bureau of Mines published in 1921. It explained in detail how the material could

be extracted and used. He continued the produce the isotope for at least nine years after that, honing the technique and sending the finished product back to Miner. "The laboratory for refining of mesothorium has

now been in operation for twelve years." Schlundt wrote in 1931. "More than 3,600 milligrams of high grade mesothorium...have been produced mainly by graduate students working under the direction of Dr. G.F. Breckenridge and the writer." These 3,600 milligrams had a market price

of between \$216,000 and \$360,000 at the time, according to prices Schlundt quoted in his letters. In today's dollars, this would be about \$3 billion to \$4 billion, depending on the years for which inflation is calculated.

Most of the radium-228 returned to Welsbach, sent in increments spread out over years. When Schlundt left MU in 1921 to study for a year at the Cavendish Laboratory in England, his students kept up his work under Breckenridge's supervision. Breckenridge, another MU chemistry professor, left the university shortly before Schlundt returned in 1922.

Schlundt the businessman

Throughout the 1920s, the refinery in Pickard Hall churned out radium-228. Schlundt's letters indicate that Welsbach paid him as a consultant during that time.

In a letter he wrote in 1922 to a company that wanted to enlist his services, he said, "At the present time I am serving in a consulting capacity for the Welsbach Company on Mesothorium. Since the process for treating mesothorium and radium are nearly identical, it seems to me that I should not accept any other consulting work with a competing firm unless I obtained consent of the Welsbach Company.³

But in a few months, he wrote, he might be able to work something out with the second company. "My contract with Welsbach expires early in July," Schlundt wrote. "I shall then be free I hope and will stand ready to cooperate with you and





Former MU chemistry department chair Herman Schlundt poses in this undated photograph. taken sometime between 1902 and 1937.



I spent last summer in Colorado in the U.S. Bureau of Mines and came in touch with the experiments in progress there on the separation of radium from the low grade ores of western Colorado."

HERMAN SCHLUNDT IN A 1914 LETTER

radium bromide salts.

your firm to the best of my ability. This kind of arrangement seems to have been against the chemistry department's rules at the

time. A set of university policies the department recommended to the university's executive boards in 1916 forbade faculty from using university laboratories, equipment or materials for commercial activities without the consent of the dean or department chair. It could have helped that Schlundt was chair

of the chemistry department from 1910 until his death in 1937, according to a history of the chemistry department by former MU chemistry professor Dorothy

Nightingale. Plus, he doled out favors with the the same enthusiasm as he doled out radium. He was constantly finding someone a job, answering those who wrote to him with chemistry questions and offering advice on how to improve chemistry education at rural high schools and colleges. He was, in many ways, a model citizen.

Schlundt's business arrangements also earned the chemistry department free equipment and supplies. When he wrote in the early 1930s to then-president Walter Williams asking for equipment, Schlundt mentioned he had sought financial support from private industry for most of his work.

"For nearly twenty five years the Chemistry department has conducted research work in the new field of science, which has revolutionized the theories of the structure of matter," Schlundt wrote. "By far the major parts of the expensive material needed for these investigations has been loaned or donated by private individuals or firms."

Friendship pays off

His friendship with Harlan Miner of Welsbach also allowed Schlundt to make high-profile donations of thorium-228 to prestigious laboratories in the U.S. and Europe. He had discovered that he could draw this bonus material, then called radiothorium, from the same lantern mantle sludge he used to refine radium-228.

"To secure the radiothorium we must have the good will of Dr. Miner of the Welsbach Company, as it is through his firm that we get our supply of mesothorium," he wrote to the would-be sponsor in 1922

One of his donations was sent to the Paris labo-



PLATE NO. 1. LEACHING POTS, BOXED, AND SUCTION FILTERS. Courtesy of UNIVERSITY ARCHIVES (C:6/17/1) In Pickard Hall, a researcher maneuvers a drum used in the extraction of radium from carnotite ore. The first phase of the procedure involved mixing the ore with acid in this drum. Howard Barker and Herman Schlundt, a former MU chemistry

professor, published this photo along with the results of their

testing water from

one of the university's

radioactivity. It tested positive. Schlundt then expanded his search for radiation to hot springs and aquifers all across the country.

Park for the U.S. **Geological Survey**.

One of Schlundt's early experiments was

deep wells for

From 1906 to 1908. he tested the springs in Yellowstone National

physics and chemistry for her work on radioactivity. Curie wrote a thank you letter to Schlundt through Miner. "I will soon receive the preparation of Radiotho-

ratory of Marie Curie, who won a Nobel Prize in

rium, which Mr. Pr. Schlundt offered to prepare in his laboratory," Curie wrote to Miner in French. "I would ask you to thank Mr. Pr. Schlundt on my behalf." Curie's letter came in 1930 and marks the height

of Schlundt's prominence in his field. But like many researchers who dealt too carelessly with radioactive materials, his work took a deadly toll on his health.

In the decades of research and industrial refining at MU, Schlundt, his colleagues and his students had all been exposed to to dangerous levels of radiation.

From cure to killer

Radiation poisoning became a national health scare in the early 1930s, when press coverage of a lawsuit against a chemical company by factory workers who had been exposed to radium developed into the "Radium Girls" scandal. One of radium's uses was in painting watch dials and gun sights — the radioactivity would

make paint glow in the dark. The women had worked at the United States Radium Corp.'s factory in East Orange, N.J., in the late 1910s and early 1920s, where they painted watch dials. Many of them would lick the tips of their tinv brushes into fine points. ingesting toxic levels of radium, according to a report Schlundt wrote for the U.S. Public Health Service.

Papers warning of the dangers of handling radium began appearing in scientific journals in the 1910s and early 1920s. Many of them were published in the French Journal of Radiology, Electrology and Nuclear Medicine. One paper by Dr. Thomas Ordway titled "Occupational Injuries Due to Radium" appeared in the Journal of the American Medical Association in 1916. Radium Corp. collected many of these papers and republished their abstracts in a book dated 1922.

About 12 years after some of the dial plant workers were exposed, they became deathly ill. The radium they had accidentally eaten had lodged in their bones. For some of them,

it caused their jaws to rot away. According to environmental historians Bill Kovarik and Mark Neuzil, five of the dial plant workers sued Radium Corp. for \$250,000 each in 1927. Newspapers went from trumpeting the health benefits of radium to playing up the pain and suffering of the dying women.

"Most of the news media dove in with a mixture of sensationalism and muckraking that accelerated and expanded the controversy," Kovarik and Neuzil noted.

The publicity put pressure on Radium Corp. to settle. Later that year, the company agreed to pay each woman \$10,000 and \$600 every year for the rest of their short lives. The company would also pay their future medical expenses.

Radium Corp. had a marked connection to Schlundt. The company had loaned him ore during his early experiments with Barker. By the time of the scandal, Barker had become a vice president of the company

Before the factory workers' case went to trial, Barker and Schlundt were writing back and forth, with Schlundt offering his opinion on dosage lev-

In 1928, the Surgeon General convened a meeting with officials from the Public Health Service and the National Consumers League. Those at the conference agreed that two committees should be set up to investigate workplace practices with radium.

Schlundt was placed on one of the committees to do research on some of the women. He traveled to

New York about 1931 to test two women who had worked at the Radium Corp. factory in the late 1910s, and he published a report of his work there. "The two girls, who after a lapse of nearly 12 vears, are still radioactive, present cases of more than passing interest, inasmuch as they stand as striking examples of the tolerance of living persons for radium," he wrote.

Declining health

Schlundt proved he was willing to subject himself to the same risks he had exposed others to. He drank water spiked with a known dose of radium to find out how quickly it would stop showing up in his urine. "As high as 91 percent of the radium taken was eliminated during the first four days after drinking the radioactive water," he wrote in his report to the Public Health Service.

When he returned to Columbia, he turned his attention to health hazards. In his letter to Walter Williams, he described some of his health studies in the early 1930s. He published a paper in the Journal of Industrial Hygiene in 1931 titled, "Dangers in Refining Radioactive Substances."

"Our first study of this problem indicates that the refining operations may be conducted without hazard when proper safety measures which we have introduced are followed by the workers in the refining laboratory," he wrote.

By then, Schlundt had begun to suffer health problems probably related to his research. In a 1933 letter, he complained of recurring bouts of sleeping sickness. He developed severe encephalitis that year and spent much of the 1933-1934 school year in the hospital, according to Nightingale's history. When he returned in 1934, he had to cut his hours.

He died of uremic poisoning, a result of kidney failure, in 1937. He was 68.

Schlundt's radioactive legacy

No mention of Schlundt's industrial refinery appeared in a two-part obituary published in the Missourian in 1937. It referred to him as an authority on radioactivity who also found time to be involved in scientific fraternities, professional chemistry societies, the Columbia Kiwanis Club and the Columbia Red Cross. It noted his connection to Curie.

It would be years before the university understood the radioactive mess he left behind. Peter Ashbrook, former director of the university's Environmental Health and Safety department explained the situation to the Nuclear Regulatory Commission in a hearing

in June 2011. Ashbrook told the commission the university has known about Pickard's radiation since the late 1970s, when staff surveyed the building and removed contaminants "where it was easy to do so."

Some of the radioactive left overs were behind walls and under floors, making them hard to reach. They placed metal shields over some spots and

restricted access to the attic and basement. They placed dosimeters, which calculate the dose of radiation a person receives when exposed. "We are not aware of anyone having been harmed by radiation in Pickard Hall," Ashbrook

Questions still remain about what if any harm might have come to Schlundt's students, the ones who faced direct exposure to more concentrated radioactive materials. These men would have been in their 20s in the 1920s. Their lives were not nearly as well-documented as Schlundt's.

Despite the danger he unwittingly posed to his students, he won the admiration of MU's student body. "The University freshman in need of a firmer grasp on the life about him found homesickness or discouragement considerable lessened through the words of Dr. Schlundt," the obituary stated

The final paragraphs of Schlundt's obituary reveal his greatest talent, the reason he was able to accomplish everything he was in his life. "He possessed an unusually retentive memory

for names and faces. His greatest hobby was people."

Supervising editor is Jeanne Abbott: abbottjm@ missouri.edu, 882-5741.

1927 Schlundt returns to Columbia and continues refining Welsbach Co.'s waste in his laboratory until around 1930. Welsbach pays Schlundt as a consultant

The price of radium goes from \$115 to \$120 a milligram to \$70 a milligram after a new source of ore is discovered in the Belgian Congo. American ore mining stops. Schlundt focuses his refining process on Welsbach's

1922-1930 Graduate students working under Schlundt's direction churr out 3.600 milligrams of radium-228. Most of it returns to Welsbach Co.

Former workers at U.S.

1928

work in a 1923 research bulletin.

Radium Corp.'s watch dial plant in Orange, N.J., sue the company after experiencing radium poisoning. The workers were painting watch dials with radium-infused glow-in-the-dark paint. Many of them would lick the tips of their tinv brushes into fine points, ingesting toxic levels of radium

Muckraking newspape

'Radium Girls' scandal

Radium Corp. settles

workers, paying them

\$10,000 and \$600 a

year for the rest of their

short lives. The Surgeon

General places Schlund

investigate the hazards

of working with radium.

on a committee to

with the dial plant

coverage feeds the

ca. 1930 1933 Schlundt makes high-profile donations of thorium-228 to several majo laboratories in the U.S. and Europe. One of them is Marie Curie's in France, Curie writes to Schlundt through Miner thanking him.

ca. 1931

had worked at

water spiked with a

known quantity of

radium to see wher

it stops showing up

in his urine.





Late 1970searly 2000s The university continues to

monitor radiation in the building and attempts to restrict access to certain areas in its basemen and attics. Ashbrook told the commission in 2011 that over the years, his office "took additional steps to reduce exposures even more to be...as

low as reasonably achievable.

2007

New regulations take effect

Regulatory Commission to

regulate sites with naturally

that require the Nuclear

occurring radioactive

materials. They requires

the commission's list of

these sites to be taken off

sites to monitor. The official

term is "decommissioning."

2009 MU notifies the commission that the new regulations apply

asks MU to

cleanup plan.

May 2013 MU announces Pickard's closure. Pickard must be empty by the end of December so a new round of testing can be



2011 The university asks the commission for an indefinite extension on the clean-up timeline. The commission denies the

request.

1937 Schlundt travels to Schlund New York to test dies of two women who uremic poisoning the Radium Corp. a result factory in the late of kidney 1910s. He drinks failure.

Late 1970s

University officials become aware of lingering radiation in the building. At a hearing in 2011, former director of MU's Office of Environmental Health and Safety Peter Ashbrook told the Nuclear Regulatory Commission that after discovering the radiation, MU staff removed contamination in areas within reach and place shields over patches of radioactivity. Some contamination remains behind walls and beneath floors. They place exposure-measuring dosimeters in the building, which return low readings.

We are not aware of anyone

having been harmed by

said

radiation in Pickard Hall." PETER ASHBROOK Former director of the Environmental Health and Safety department at MU